# 103. Internal Nucleophilic Termination in Acid-Mediated Polyene Cyclisations: Synthetic Access to Methyl Homologs of ( $\pm$ )-Ambrox ${ }^{(\mathbb{1})}$ and Its Diastereoisomers 

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#### Abstract

Treatment of ten monocyclic dienols 8-11 with an excess of fluorosulfonic acid in 2-nitropropane at $-90^{\circ}$ afforded diastereoisomeric mixtures of racemic tricyclic ethers 12-14 in 81-91\% yield (see Tables 1 and 2). These transformations represent further examples of biomimetic acid-mediated cyclisations in which an OH group serves as the internal nucleophilic terminator. A non-synchronous process is postulated, and the examples described strongly re-inforce our working mechanistic hypothesis, whereby the stereochemical course of cyclisation is directed by the orientation of the side chain vicinal to the intermediate cyclohexyl cation (see Schemes 4 and 5). It is also demonstrated that the efficiency of this process is independent of the nature of the OH group, which may be primary, secondary, or tertiary. In addition, the organoleptic properties of 12-14, Me homologs of known odorants such as $A m b r o x{ }^{39}((-)-3 a)$ and its diastereoisomers, are briefly discussed.


Introduction. - Recently, we described an efficient biomimetic access to organoleptically active tricyclic ethers, by treatment of appropriately substituted 6-(cyclohex-enyl)hex-3-en-1-ols with an excess of fluorosulfonic acid in 2-nitropropane at $-90^{\circ}[1]$. These kinetically controlled cyclisations ${ }^{1}$ ) were shown to proceed stereospecifically via protonation of the cyclohexenyl bond, followed by ring closure involving equatorial $\mathrm{C}-\mathrm{C}$ bond formation with concomitant internal nucleophilic termination by anti-addition of the OH group across the $\mathrm{C}(3)=\mathrm{C}(4)$ bond. For example, allowing for partial acidcatalysed isomerisation of the $\mathrm{C}(3)=\mathrm{C}(4)$ bond, $(E)$ - and $(Z)-1$ selectively afford 3a and $\mathbf{3 b}$, whereas $(E)$ - and ( $Z$ )-2 preferentially generate $\mathbf{3 c}$ and 3 d , via a favoured chair conformation of the nascent cyclohexane ring (see Scheme 1). It was also established that conformational inversion of the six-membered ring is slower than ring closure, the orientation of the $\mathrm{C}\left(1^{\prime}\right)$-side chain vicinal to the cyclohexyl cation thus directing the stereochemical course of cyclisation. For $\mathbf{1 \rightarrow 3 a} / \mathbf{3} \mathbf{b}$, this orientation is determined by stereoselective axial protonation of the tetrasubstituted cyclohexenyl bond, and ensures a trans- $A / B$ ring junction; in contrast, the predominantly pseudoaxial orientation of the side chain in 2 results in the generation of $\mathbf{3 c} / \mathbf{3 d}$ with a cis- $A / B$ ring junction. In continuation of our studies, we now report the preparation and acid-mediated cyclisation of ten Me-substituted homologs of 1 and 2. During this investigation, we planned to address two questions. Firstly, can the proposed mechanistic hypothesis predict the stereochemical outcome of cyclisation for substrates substituted at C( $5^{\prime}$ )? Secondly, is

[^0]Scheme 1. Acid-Mediated Cyclisations of (E)-and (Z)-1, (E)- and (Z)-2: Major Reaction Pathways

ring closure compatible with substitution at $\mathrm{C}(2)$, a situation in which internal termination is necessarily effected by a secondary or tertiary OH group? In addition, from a perfumistic viewpoint, it was of interest to discover the organoleptic properties of the racemic Me homologs ${ }^{2}$ ) of odorants such as $\mathbf{3 a}^{\mathbf{3}}$ ) and its $\mathrm{C}(9 \mathrm{~b})$-epimer, $\mathbf{3 b}^{4}$ ).

Results and Discussion. - Stereochemically pure samples of the ten alcohols, $(E)$ - and ( $Z$ )-8, cis -9 , trans $-9,10$, and 11, were synthesised using standard methodology (see Schemes 2 and 3). Thus, pure samples of $\beta$-irone, cis- $\alpha$-irone, and trans- $\alpha$-irone, obtained by fractional distillation of a commercial mixtures ${ }^{5}$, were catalytically monohydrogenated to ketones 4, cis-5, and trans-5, respectively, in $80-90 \%$ yield. Wittig reaction using the ylide derived from (3-hydroxypropyl)triphenylphosphonium bromide [10] then afforded 8, cis -9 , and trans -9 as $1: 1$ diastereoisomeric mixtures, in $40-50 \%$ yield. The known $\beta, \gamma$-unsaturated acid $6((E) /(Z) 2: 1)[11]$ was treated with MeLi to give ketone 7 $((E) /(Z) 2: 1,66 \%)$; subsequent hydride reduction afforded $10((E) /(Z) 2: 1,94 \%)$,

[^1]Scheme 2



i) $\mathrm{H}_{2}$, Raney- Ni , EtOH , r.t. (yield: $80-90 \%$ ); ii) $\left[\mathrm{Ph}_{3} \mathrm{P}\left(\mathrm{CH}_{2}\right)_{3} \mathrm{OH}\right]^{\oplus} \mathrm{Br}^{\ominus}$, BuLi (2 mol-equiv.), THF, r.t. (yield: 40-50\%).

Scheme 3


i) $\mathrm{MeLi}\left(2.5\right.$ mol-equiv.), $\mathrm{Et}_{2} \mathrm{O},-60^{\circ} \rightarrow$ r.t.;
ii) $\mathrm{LiAlH}_{4}, \mathrm{Et}_{2} \mathrm{O},-30^{\circ}$;
iii) MeMgI ( 2.9 mol-equiv.), $\mathrm{Et}_{2} \mathrm{O}, 5^{\circ}$.

$11((E) /(Z) 2: 1)$
$\left.{ }^{\text {a }}\right) 11((E) /(Z) 2: 1,4 \%)$ also isolated, see Exper.Part.
whereas treatment with MeMgI furnished $11((E) /(Z) 2: 1,95 \%)$. Separation of the $(E) /(Z)$-diastereoisomeric mixtures 8 - 11 was readily effected by column chromatography.

Standard cyclisation conditions involved treatment of each alcohol with a ten-fold excess of fluorosulfonic acid in 2-nitropropane at $-90^{\circ}$. Subsequent neutralisation with aqueous $\mathrm{NaHCO}_{3}$ solution, followed by an extractive workup, afforded mixtures of tricyclic ethers ${ }^{6}$ ) in $80-90 \%$ yield; the product distributions are presented in Tables 1 and 2. Structural identification of the products was effected either by preparative GC fol-

[^2]Table 1. Acid-Mediated Cyclisations of 8, cis-9, and trans-9

| Entry | Substrate ${ }^{\text {a }}$ ) | Product distribution (yield [\%]) ${ }^{\text {b }}$ ) |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 12a | 12b | 12c | 12d | 12e | 12 f |
| 1 | (E)-8 | 48 | 35 | - | - | - | 3 |
| 2 | ( $Z$ )-8 | - | 84 | - | 3 | - | 4 |
| 3 | (E)-cis-9 | 48 | 39 | - | - | - | 2 |
| 4 | (Z)-cis-9 | 2 | 77 | - | 3 | - | 4 |
| 5 | (E)-trans-9 | - | - | 55 | 3 | 27 | - |
| 6 | ( $Z$ )-trans-9 | - | 1 | 8 | 57 | 4 | 16 |

${ }^{4}$ ) Reaction conditions: substrate ( 1 mmol ), $\mathrm{FSO}_{3} \mathrm{H}(10 \mathrm{mmol}), 2$-nitropropane $\left.(15 \mathrm{ml}),-90^{\circ} .{ }^{b}\right)$ GC Analysis of distilled product after workup.


Table 2. Acid-Mediated Cyclisations of $\mathbf{1 0}$ and 11

| Entry | Substrate ${ }^{\text {a }}$ ) | Product distribution (yield [\%]) ${ }^{\text {b }}$ ) |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 13a ${ }^{\text {c }}$ ) | 13b ${ }^{\text {c }}$ ) | 14a | 14b |
| 1 | (E)-10 | 55 | 26 |  |  |
| 2 | (Z)-10 | - | 82 |  |  |
| 3 | (E)-11 |  |  | 63 | 23 |
| 4 | (Z)-11 |  |  | - | 85 |

${ }^{\text {a }}$ ) Reaction conditions: see Table 1. ${ }^{\text {b }}$ ) GC Analysis of distilled product after workup; minor amounts (5-8\%) of cis-decalin products not characterised. ${ }^{\text { }}$ ) Between $1: 1$ and $2: 1$ diastereoisomeric mixture at $\mathrm{C}(2)$.


13a


13 b


14 a

$14 b$
lowed by full spectral characterisation (viz. 12a-f) or, where separation of diastereoisomers was not possible, GC/MS and NMR analysis of the mixtures, corroborated by comparison with authentic or closely analogous compounds was sufficient (viz. 13a,b [3], 14a,b [4]).

Cyclisation of ( $E$ )-8 afforded two major products, 12a ( $48 \%$ ) and 12b ( $35 \%$ ), together with a minor amount of $\mathbf{1 2 f}(\mathbf{3} \%$; Table 1, Entry 1); in contrast, (Z)-8 gave almost exclusively $\mathbf{1 2 b}(84 \%)$ accompanied by $\mathbf{1 2 d}(\mathbf{3} \%$ ) and $12 f(4 \%$; Table 1, Entry 2 ). In close analogy with the cyclisations of 1 and 2 (see Scheme 1), these results can be explained by a non-synchronous pathway involving stereoselective axial protonation of the tetrasubstituted cyclohexenyl bond to carbocations $(E)$ - and $(Z)$-I in which the $\mathrm{C}\left(5^{\prime}\right)-\mathrm{Me}$ group

Scheme 4. Acid-Mediated Cyclisations of (E)-and (Z)-8, (E)- and (Z)-cis-9, and (E)-and (Z)-trans-9



occupies an energetically favourable pseudoequatorial orientation, followed by cyclisation to products possessing a trans- $A / B$ ring junction (see Scheme 4). Also analogously, cyclisation of $(E)$-8 to $\mathbf{1 2 a}$ completes with its isomerisation to $(Z)-\mathbf{8}$, whose ring closure to 12b is more rapid. Mechanistically, the cyclisations of ( $E$ )- and ( $Z$ )-cis-9 (Table 1, Entries 3 and 4) are important, as they provide conclusive evidence that, when the cyclohexenyl bond is trisubstituted, the conformation of the substrate determines the stereochemical course of cyclisation. Because the cis-relationship of the $\mathrm{C}\left(1^{\prime}\right)$-side chain and the $\mathrm{C}\left(5^{\prime}\right)-\mathrm{Me}$ group forces these substituents into pseudoequatorial orientations, protonation is predicted to again lead to cyclohexyl cations, $(E)$ - and $(Z)$-I, and thus the same product distribution which results from $(E)$ - and $(Z)-\mathbf{8}$, respectively. Within experimental error, this is indeed the case. Thus, cyclisation of (E)-cis-9 afforded 12a (48\%) and 12b ( $39 \%$ ) as major products, with minor amounts of $\mathbf{1 2 f}(2 \%)$; (Z)-cis-9 gave a mixture
containing 12b (77\%) with small quantities of 12a (2\%), 12d (3\%), and 12f (4\%). For $(E)$ - and ( $Z$ )-trans-9, the lowest-energy conformer has the $\mathrm{C}\left(1^{\prime}\right)$-side chain pseudoaxial with the $\mathrm{C}\left(5^{\prime}\right)$-Me group pseudoequatorial'), and thus their cyclisations mainly lead to products with a cis- $A / B$ ring junction. Protonation of $(E)$-trans-9 to carbocations ( $E$ )-II (chair conformation) or $(E)-\mathrm{II}^{\prime}$ (skew-boat conformation) ${ }^{8}$ ), and subsequent cyclisation, furnished 12c (55\%) and 12e ( $27 \%$ ), respectively (Table 1, Entry 5). Similarly, the major products from the cyclisation of ( $Z$ )-trans-9 were 12d ( $57 \%$ ) and $\mathbf{1 2 f}$ ( $16 \%$; Table 1, Entry 6), via carbocations ( $Z$ )-II (chair conformation) and ( $Z$ )-II' (skew-boat conformation) ${ }^{7}$ ), respectively (see Scheme 4).

Cyclisation of ( $E$ )-10 afforded two main products, 13a ( $55 \%$ ) and 13b ( $26 \%$ ), whereas ( $Z$ )-10 gave exclusively $\mathbf{1 3 b}(82 \%$; Table 2, Entries 1 and 2 ). It is worth noting that, not unexpectedly, both 13a and 13b are diastereoisomeric mixtures at $\mathrm{C}(2)$, the cyclisation proceeding with almost no stereochemical control at this centre ${ }^{9}$ ). Similarly, cyclisation of $(E)$-11 afforded a mixture $14 \mathrm{a}(63 \%) / \mathbf{1 4 b}(23 \%)$, whereas ( $Z$ )-11 stereospecifically furnished $\mathbf{1 4 b}$ ( $85 \%$; Table 2, Entries 3 and 4). These results thus conclusively demonstrate that cyclisation proceeds efficiently, when the internal terminating group is either a secondary or a tertiary alcohol, and that the same non-synchronous process is operative (see Scheme 5). As observed previously (vide supra), it is

Scheme 5. Acid-Mediated Cyclisations of (E)- and (Z)-10, (E)-and (Z)-11


[^3]again evident that the $(Z)$-isomers of $\mathbf{1 0}$ and $\mathbf{1 1}$ cyclise faster than the corresponding $(E)$-isomers, and that partial isomerisation of the $C(4)=C(5)$ band in the substrate alcohol competes with the cyclisation process ${ }^{10}$ ).

A comparison of the olfactive properties of the six diastereoisomeric tricyclic ethers, 12a-f, was possible due to their separation and isolation by preparative GC. In accordance with the organoleptic properties of their C(7)-demethylated analogues [8] [13], the two trans- $A / B$ ring-junction isomers 12a,b have strong amber, woody odours; in contrast, the four cis- $A / B$ ring-junction isomers $12 \mathbf{c}-\mathbf{f}$ exhibited woody, resinous, perspiration notes with only weak amber undertones. Diastereoisomeric mixtures 13a,b possessed typical amber, woody notes which are nonetheless weaker than the C(2)-demethylated analogues. The corresponding dimethylated homologs $\mathbf{1 4 a}$ and $\mathbf{1 4 b}$ were odourless.

## Experimental Part

(with the collaboration of O. Barbuzzi, H. Pamingle, and P. Sonnay)

1. General. See [1].
2. Materials. A commercial mixture (Givaudan-Roure) of (E)-4-(2', $5^{\prime}, \sigma^{\prime}, \sigma^{\prime}$-tetramethylcyclohex- $I^{\prime}-$ enyl)but-3-en-2-one ( $\beta$-irone, 4\%), (E)-4-[(I'RS, $5^{\prime}$ RS)-2', $5^{\prime}, 6^{\prime}, 6^{\prime}$-tetramethylcyclohex- $2^{\prime}$-enyl]but-3-en-2-one (cis- $\alpha$-irone, $51 \%$ ), and ( E )-4-[( $I^{\prime} \mathrm{RS}, 5^{\prime} \mathrm{SR}$ )-2', $5^{\prime}, 6^{\prime}, 6^{\prime}$-tetramethylcyclohex-2'-enyl]but-3-en-2-one (trans- $\alpha$-irone, $45 \%$ ) was fractionally distilled i.v. to obtain anal. pure samples (distillation order: trans- $\alpha$-irone, cis- $\alpha$-irone, and $\beta$-irone).
3. Preparation of Ketones 4, cis-5, and trans-5. 4-( $2^{\prime}, 5^{\prime}, 6^{\prime}, 6^{\prime}$-Tetramethylcyclohex- $I^{\prime}$-enyl)butan-2-one (4), ( $1^{\prime} \mathrm{RS}, 5^{\prime} \mathrm{RS}$ )-4-( $2^{\prime}, 5^{\prime}, 6^{\prime}, 6^{\prime}$-tetramethylcyclohex-2'-enyl) butan-2-one (cis-5), and ( $1^{\prime} \mathrm{RS}, 5^{\prime} \mathrm{SR}$ )-4-( $2^{\prime}, 5^{\prime}, 6^{\prime}, 6^{\prime}$-tetra-methylcyclohex-2'-enyl)butan-2-one (trans-5) were prepared by catalytic monohydrogenation (Raney- $\mathrm{Ni} / \mathrm{EtOH}$ ) of $\beta$-irone, cis- $\alpha$-irone, and trans- $\alpha$-irone, respectively; purification (purity $\geq 99 \%$ ) was effected by fractional distillation i.v.

Data of 4: b.p. $88-92^{\circ} / 0.4$ Torr; identical ( $t_{\mathrm{R}}, \mathrm{MS}$ ) with an authentic sample [14]. ${ }^{13} \mathrm{C}$-NMR: $208.9(s) ; 136.3$ $(s) ; 127.5(s) ; 44.7(t) ; 39.4(d) ; 38.3(s) ; 31.7(t) ; 29.7(q) ; 27.3(t) ; 26.7(q) ; 22.7(t) ; 21.6(q) ; 19.9(q) ; 16.6(q)$.

Data of cis-5: b.p. 60-61 $/ 0.02$ Torr; identical ( $t_{\mathrm{R}}, \mathrm{MS}$ ) with an authentic sample [14].
Data of trans-5: b.p. $52^{\circ} / 0.01 \mathrm{Torr}$; identical ( $t_{\mathrm{R}}, \mathrm{MS}$ ) with an authentic sample [14].
4. 5-Methyl-7-( $2^{\prime}, 6^{\prime}, 6^{\prime}$-trimethylcyclohex- $I^{\prime}$-enyl hept-4-en-2-one ( $7 ;(E / Z)$ 2:1). MeLi ( 92 ml of a ca. 1.6 m soln. in $\mathrm{Et}_{2} \mathrm{O}$; Fluka; 0.15 mol ) was added dropwise within l h to a stirred soln. of 4 -methyl- $6-\left(2^{\prime}, 6^{\prime}, 6^{\prime}-\right.$ trimethyl-cyclohex-I'-enyl) hex-3-enoic acid $(6 ;(E) /(Z) 2: 1[11] ; 12 \mathrm{~g}, 0.048 \mathrm{~mol})$ in $\mathrm{Et}_{2} \mathrm{O}(200 \mathrm{ml})$ at $-60^{\circ}$ under $\mathrm{N}_{2}$. After 2 h at $-60^{\circ}$, the mixture was allowed to attain r.t. during 2 h and stirred at r.t. for 18 h . The mixture was then poured cautiously into cold $10 \%$ aq. $\mathrm{NH}_{4} \mathrm{Cl}$ soln. ( 300 ml ) and saturated with NaCl . Separation of the org. phase and extraction of the aq. phase $\left(\mathrm{Et}_{2} \mathrm{O}\right)$ afforded a combined org. phase which was successively washed with aq. 2 N NaOH soln. and $\mathrm{H}_{2} \mathrm{O}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated i.v. Chromatography ( $\mathrm{SiO}_{2}(360 \mathrm{~g})$, toluene/AcOEt 9:1) followed by distillation i.v. afforded $7((E) /(Z) 2: 1)$ as a pale yellow oil ( $7.8 \mathrm{~g}, 66 \%$ ). TLC (toluene/AcOEt 9:1) $R_{\mathrm{f}} 0.47$.

Data of ( E$)-7$ : ${ }^{1} \mathrm{H}-\mathrm{NMR}: 1.00(\mathrm{~s}, 6 \mathrm{H}) ; 1.42(\mathrm{~m}, 2 \mathrm{H}) ; 1.58(\mathrm{~m}, 2 \mathrm{H}) ; 1.60(\mathrm{~s}, 3 \mathrm{H}) ; 1.68(\mathrm{~s}, 3 \mathrm{H}) ; 1.91(t, J=7$, $2 \mathrm{H}) ; 2.08(4 \mathrm{H}) ; 2.15(s, 3 \mathrm{H}) ; 3.13(d, J=7,2 \mathrm{H}) ; 5.36$ (br. $t, J=7,1 \mathrm{H}) . \mathrm{MS}: 248\left(1, M^{+}\right), 190(5), 137(88), 95$ (100), 81 ( 80 ), 43 (75).

Data of (Z)-7: ${ }^{1} \mathrm{H}$-NMR: $1.01(s, 6 \mathrm{H}) ; 1.42(\mathrm{~m}, 2 \mathrm{H}) ; 1.58(m, 2 \mathrm{H}) ; 1.64(s, 3 \mathrm{H}) ; 1.81(s, 3 \mathrm{H}) ; 1.92(t, J=7$, $2 \mathrm{H}) ; 2.04(4 \mathrm{H}) ; 2.15(s, 3 \mathrm{H}) ; 3.15(d, J=7,2 \mathrm{H}) ; 5.29$ (br. $t, J=7,1 \mathrm{H}) . \mathrm{MS}: 248\left(2, M^{+}\right), 190(6), 137(74), 95$ (100), 81 (84), 43 (78).

Also isolated was $11((E) /(Z) 2: 1 ; 0.48 \mathrm{~g}, 4 \%)$. TLC (toluene/AcOEt 9:1): $R_{\mathrm{f}} 0.25$. Identical ( $t_{\mathrm{R}}, \mathrm{MS}$ ) with an authentic sample (vide infra).

[^4]5. Preparation of (E)- and (Z)-8, (E)- and (Z)-cis-9, (E)- and (Z)-trans-9. (E)- and (Z)-4-methyl-6(2', $5^{\prime}, 6^{\prime}, \sigma^{\prime}$-tetramethylcyclohex- $1^{\prime}$-enyl) hex-3-en-1-ol ( $(E)$ - and (Z)-8), (E)- and (Z)-4-methyl-6-[( $1^{\prime}$ RS, $5^{\prime}$ RS)$2^{\prime}, 5^{\prime}, 6^{\prime}, 6^{\prime}$-tetramethylcyclohex-2'-enyllhex-3-en-l-ol ( $(E)$ - and ( $Z$ )-cis-9), and (E)- and (Z)-4-methyl-6[ ( $\left.l^{\prime} \mathrm{RS}, 5^{\prime} \mathrm{SR}\right)-2^{\prime}, 5^{\prime}, 6^{\prime}, 6^{\prime}$-tetramethylcyclohex-2'-enyl]hex-3-en-1-ol ( $(E)$ - and $(Z)$-trans-9) were prepared from 4, cis-5 and trans-5, respectively, using a standard Wittig procedure [10]. In each case, a ca. 1:1 ( $E / Z$ )-diastereoisomeric mixture of the product alcohol ( $2-3 \mathrm{~g}$ ) was obtained ( $\mathrm{ca} .40-50 \%$ yield based on recovered, unreacted starting ketone). Chromatographic purification ( $\mathrm{SiO}_{2}(360 \mathrm{~g})$, toluene/AcOEt 19:1; $(E)$-isomer more polar than $(Z)$-isomer) followed by Kugelrohr distillation ( $140-160^{\circ} / 0.02 \mathrm{Torr}$ ) afforded pure samples of each diastereoisomer (ca. $0.5-1 \mathrm{~g}$, purity ( GC ) $\geq 99 \%$ ).

Data of (E)-8: ${ }^{1} \mathrm{H}$-NMR: $0.84(s, 3 \mathrm{H}) ; 0.88(d, J=6.5,3 \mathrm{H}) ; 1.02(s, 3 \mathrm{H}) ; 1.60(\mathrm{~s}, 3 \mathrm{H}) ; 1.68(\mathrm{~s}, 3 \mathrm{H}) ; 3.62(\mathrm{~m}$, $2 \mathrm{H}) ; 5.17(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ : see Table 3. MS: $250\left(2, \mathrm{M}^{+}\right), 151$ (36), $109(46), 95(100), 81$ (39), 67 (28), 55 (27), 41 (28).

Data of (Z)-8: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.85(\mathrm{~s}, 3 \mathrm{H}) ; 0.88(d, J=6.5,3 \mathrm{H}) ; 1.14(\mathrm{~s}, 3 \mathrm{H}) ; 1.65(\mathrm{~s}, 3 \mathrm{H}) ; 1.78(\mathrm{~s}, 3 \mathrm{H}) ; 3.63(\mathrm{~m}$, $2 \mathrm{H}) ; 5.11(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$-NMR: see Table 3. MS: $250\left(1, \mathrm{M}^{+}\right), 151(51), 135(10), 121(12), 109(57), 95(100), 81(40)$, 67 (31), 55 (23), 41 (29).

Data of (E)-cis-9: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.63(s, 3 \mathrm{H}) ; 0.84(d, J=7,3 \mathrm{H}) ; 0.89(s, 3 \mathrm{H}) ; 1.66(s, 3 \mathrm{H}) ; 1.70(\mathrm{~s}, 3 \mathrm{H}) ; 3.62$ $(t, J=6,2 \mathrm{H}) ; 5.16(t, J=7,1 \mathrm{H}) ; 5.32(m, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ : see Table 3. MS: $250\left(0, M^{+}\right), 150(71), 135(60), 121$ (28), 107 (37), 95 (30), 81 (100), 69 (27), 55 (33), 41 (35).

Data of (Z)-cis-9: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.64(s, 3 \mathrm{H}) ; 0.84(d, J=7,3 \mathrm{H}) ; 0.92(s, 3 \mathrm{H}) ; 1.75$ (br. $\left.s, 6 \mathrm{H}\right) ; 3.62(t, J=6$, $2 \mathrm{H}) ; 5.12(t, J=7,1 \mathrm{H}) ; 5.33$ (br. $s, 1 \mathrm{H}){ }^{13} \mathrm{C}$-NMR: see Table 3. MS: $250\left(0.5, M^{+}\right), 150(49), 135(30), 123(20)$, 107 (26), 95 (30), 81 (100), 67 (20), 55 (27), 41 (28).

Data of ( E )-trans-9: ${ }^{1} \mathrm{H}$-NMR: $0.74(s, 3 \mathrm{H}) ; 0.81(d, J=7,3 \mathrm{H}) ; 0.91(s, 3 \mathrm{H}) ; 1.65(s, 3 \mathrm{H}) ; 1.67(s, 3 \mathrm{H}) ; 1.95$ $(d, J=14,1 \mathrm{H}) ; 2.07(t, J=9,2 \mathrm{H}) ; 2.28(q, J=7,2 \mathrm{H}) ; 3.62(\mathrm{br} . s, 2 \mathrm{H}) ; 5.13(t, J=7,1 \mathrm{H}) ; 5.22$ (br. $s, 1 \mathrm{H})$. ${ }^{13} \mathrm{C}$-NMR: see Table 3. MS: $250\left(1, M^{+}\right), 150(37), 135(39), 123$ (58), 112 (33), 95 (54), 81 (100), 67 (31), 55 (38), 41 (38).

Data of (Z)-trans-9: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.75(s, 3 \mathrm{H}) ; 0.81(d, J=7,3 \mathrm{H}) ; 0.95(s, 3 \mathrm{H}) ; 1.69(s, 3 \mathrm{H}) ; 1.73(s, 3 \mathrm{H}) ; 3.62$ (br. $s, 2 \mathrm{H}) ; 5.08(t, J=7,1 \mathrm{H}) ; 5.23$ (br. $s, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$-NMR: see Table 3. MS: $250\left(2, M^{+}\right), 150(19), 135(23), 123$ (68), 112 (28), 95 (49), 81 (100), 67 (24), 55 (32), 41 (32).
6. (E)- and (Z)-5-Methyl-7-(2', $6^{\prime}, 6^{\prime}$-trimethylcyclohex-1'-enyl)hept-4-en-2-ol ( $(E)$ - and (Z)-10). A soln. of 7 $((E) /(Z) 2: 1$ (vide supra) $; 6 \mathrm{~g}, 0.023 \mathrm{~mol})$ in $\mathrm{Et}_{2} \mathrm{O}(40 \mathrm{ml})$ was added dropwise, within 10 min , to a stirred slurry of $\mathrm{LiAlH}_{4}(0.46 \mathrm{~g}, 0.012 \mathrm{~mol})$ in $\mathrm{Et}_{2} \mathrm{O}(20 \mathrm{ml})$ at $-30^{\circ}$ under $\mathrm{N}_{2}$. After a further 15 min at $-30^{\circ}, \mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{ml}), 20 \% \mathrm{aq}$. NaOH soln. $(0.5 \mathrm{ml})$, and $\mathrm{H}_{2} \mathrm{O}(3 \mathrm{ml})$ were added dropwise successively. Filtration (Hyflo), concentration of the filtrate, and distillation i.v. afforded $13((E) /(Z) 2: 1)$ as a colourless oil $\left(5.4 \mathrm{~g}, 94 \%\right.$ ), b.p. $90-98^{\circ} / 0.02 \mathrm{Torr}$, which was purified by chromatography $\left(\mathrm{SiO}_{2}(360 \mathrm{~g})\right.$, toluene/AcOEt $\left.9: 1\right)$ to furnish pure samples of $(E)-10(2.2 \mathrm{~g}$, more polar) and ( $Z$ ) -10 ( 2.0 g , less polar).

Data of (E)-10: IR ( $\mathrm{CHCl}_{3}$ ): 3540, 3390 (br.), 1440, 1370, 1350, 1250, 1100, 1036, 922. ${ }^{\mathrm{t}} \mathrm{H}-\mathrm{NMR}: 1.00(s$, $6 \mathrm{H}) ; 1.20(d, J=7,3 \mathrm{H}) ; 1.42(\mathrm{~m}, 2 \mathrm{H}) ; 1.57(\mathrm{~m}, 2 \mathrm{H}) ; 1.61(\mathrm{~s}, 3 \mathrm{H}) ; 1.69(\mathrm{~s}, 3 \mathrm{H}) ; 1.91(t, J=7,2 \mathrm{H}) ; 2.07(4 \mathrm{H}) ;$ $2.18(m, 2 \mathrm{H}) ; 3.81(\mathrm{~m}, 1 \mathrm{H}) ; 5.20(t, J=7,1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ : see Table 3. MS: $250\left(1, M^{+}\right), 137(97), 121(19), 107$ (15), 95 (100), 81 (81), 69 (25).

Data of (Z)-10: IR ( $\mathrm{CHCl}_{3}$ ): 3530, 3390 (br.), 1434, 1360, 1340, 1240, 1100, 1050, 1030, $920 .{ }^{1} \mathrm{H}-\mathrm{NMR}: 1.02$ $(s, 6 \mathrm{H}) ; 1.20(d, J=7,3 \mathrm{H}) ; 1.42(m, 2 \mathrm{H}) ; 1.57(\mathrm{~m}, 2 \mathrm{H}) ; 1.65(\mathrm{~s}, 3 \mathrm{H}) ; 1.79(\mathrm{~s}, 3 \mathrm{H}) ; 1.92(t, J=7,2 \mathrm{H}) ; 1.96-2.12$ ( 4 H$) ; 2.18(\mathrm{~m}, 2 \mathrm{H}) ; 3.79(m, 1 \mathrm{H}) ; 5.14(t, J=7,1 \mathrm{H}) .{ }^{13} \mathrm{C}$-NMR: see Table 3. MS: $250\left(0, M^{+}\right), 137(95), 121$ (17), 107 (18), 95 (100), 81 (74), 69 (28).
7. (E)- and (Z)-2,5-Dimethyl-7-(2', $6^{\prime}, 6^{\prime}-t r i m e t h y l c y c l o h e x-1^{\prime}-$ enyl) hept-4-en-2-ol ( $(E)$ - and (Z)-11). A soln. of $7((E) /(Z) 2: 1$ (vide supra) $; 4.6 \mathrm{~g}, 0.018 \mathrm{~mol})$ in $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{ml})$ was added dropwise, within 20 min , to a stirred soln. of freshly prepared $\mathrm{MeMgI}(0.053 \mathrm{~mol})$ in $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{ml})$ at $5^{\circ}$ under $\mathrm{N}_{2}$. After a further 30 min at $5^{\circ}$, the mixture was poured into stirred cold $20 \%$ aq. $\mathrm{NH}_{4} \mathrm{Cl}$ soin. $(200 \mathrm{ml})$ and saturated with NaCl . Separation of the org. phase and extraction of the aq. phase $\left(\mathrm{Et}_{2} \mathrm{O}\right)$ afforded a combined org. phase which was dried (anh. $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ), filtered, and concentrated i.u. Distillation i.u. furnished $11((E) /(Z) 2: 1)$ as a colourless oil $(4.6 \mathrm{~g}, 95 \%)$, b.p. $110-115^{\circ} / 0.1$ Torr, which was purified by chromatography $\left(\mathrm{SiO}_{2}(360 \mathrm{~g})\right.$, toluene/AcOEt 19:1) to furnish pure samples of $(E)$ - $11(1.6 \mathrm{~g})$ and $(Z)-11(1.3 \mathrm{~g})$.

Data of (E)-11: IR (CHCl $): 3420(\mathrm{br}), 1462,1380,1131,.900 .{ }^{1} \mathrm{H}-\mathrm{NMR}: 1.00(s, 6 \mathrm{H}) ; 1.22(s, 6 \mathrm{H}) ; 1.42(\mathrm{~m}$, $2 \mathrm{H}) ; 1.57(\mathrm{~m}, 2 \mathrm{H}) ; 1.61(\mathrm{~s}, 3 \mathrm{H}) ; 1.68(\mathrm{~s}, 3 \mathrm{H}) ; 1.91(t, J=7,2 \mathrm{H}) ; 2.08(4 \mathrm{H}) ; 2.19(d, J=7,2 \mathrm{H}) ; 5.28(t, J=7$, $1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}:$ see Table 3. MS: $264\left(0, M^{+}\right), 246(4), 137(100), 109(77), 95(88), 81$ (59).


| Table 4. ${ }^{13} \mathrm{C}$-NMR Assignments ( $\delta$ [ppm]) for 12a-f |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Compound | C(1) | C(2) | C(3a) | C(4) | C(5) | C(5a) | C(6) | C(7) | C(8) | C(9) | C(9a) | C(9b) | $\mathrm{C}(3 \mathrm{a})-\mathrm{Me}$ | $\mathrm{C}(6)-\mathrm{Me} \alpha_{\alpha}$ | $\mathrm{C}(6)-\mathrm{Me} \mathrm{p}_{\beta}$ | C(7)-Me | $\mathrm{C}(9 \mathrm{a})-\mathrm{Me}$ |
| $12 \mathrm{a}^{\text {a }}$ ) | 22.6 | 65.0 | 79.7 | 39.9 | 20.8 | 58.1 | $36.2^{\text {b }}$ ) | 42.8 | 27.3 | 39.8 | $36.4{ }^{\text {b }}$ ) | 60.4 | 21.1 | 29.4 | 16.1 | 16.5 | 14.9 |
| 12ba) | 29.0 | 64.1 | 80.7 | 35.9 | 20.6 | 47.8 | $36.2{ }^{\text {b }}$ ) | 43.0 | 27.4 | 38.6 | $36.4{ }^{\text {b }}$ ) | 59.1 | 27.7 | 29.6 | 16.5 | 16.5 | 22.8 |
| 12c | 24.3 | 64.4 | 78.9 | 38.8 | 23.4 | 55.4 | 36.9 | 36.5 | 28.2 | 27.6 | 38.1 | 58.5 | 20.7 | 26.4 | 31.0 | 16.1 | 29.5 |
| 12d | 29.7 | 63.3 | 80.1 | 33.6 | 22.7 | 49.1 | 37.4 | 36.1 | 28.6 | 33.9 | 36.6 | 58.3 | 27.0 | 25.9 | 31.3 | 16.2 | 29.3 |
| 12e | 23.7 | 65.2 | 80.4 | 36.9 | 21.0 | 54.2 | 34.9 | 38.2 | 26.6 | 37.6 | 36.2 | 48.8 | 24.4 | 27.3 | 30.8 | 16.0 | 24.4 |
| 12 f | 28.6 | 65.0 | 80.9 | 35.5 | 19.4 | 54.9 | 37.2 | 35.8 | 27.2 | 27.7 | 37.6 | 54.9 | 27.7 | 25.7 | 31.5 | 16.1 | 29.1 |

[^5]Data of (Z)-11: IR (CHCl ${ }_{3}$ ): 3420 (br.), 1460, 1370, 1130. ${ }^{i} \mathrm{H}-\mathrm{NMR}: 1.02(\mathrm{~s}, 6 \mathrm{H}) ; 1.22(\mathrm{~s}, 6 \mathrm{H}) ; 1.42(\mathrm{~m}, 2 \mathrm{H})$; $1.57(m, 2 H) ; 1.65(s, 3 \mathrm{H}) ; 1.80(s, 3 \mathrm{H}) ; 1.91(t, J=7,2 \mathrm{H}) ; 2.06(4 \mathrm{H}) ; 2.22(d, J=7,2 \mathrm{H}) ; 5.22(t, J=7,2 \mathrm{H})$ ${ }^{13} \mathrm{C}$-NMR: see Table 3. MS: $264\left(0, \mathrm{M}^{+}\right), 246(4), 137(100), 121$ (21), 109 (69), 95 (97), 81 (75).
8. Acid-Mediated Cyclisation of Alcohols 8-11. Preparation of (3aRS,5aSR,7SR,9aSR,9bRS)-, (3aRS, $5 a$ SR , $7 \mathrm{SR}, 9 a \mathrm{SR}, 9 b \mathrm{SR}$ )-, ( $3 a \mathrm{RS}, 5 a \mathrm{SR}, 7 \mathrm{RS}, 9 a \mathrm{RS}, 9 b \mathrm{RS}$ )-, ( $3 a \mathrm{RS}, 5 a \mathrm{SR}, 7 \mathrm{RS}, 9 a \mathrm{RS}, 9 b \mathrm{SR}$ )-, ( $3 a \mathrm{RS}, 5 a \mathrm{RS}$, 7SR, $9 a$ SR, $9 b \mathrm{RS}$ )-, and ( $3 a \mathrm{RS}, 5 a \mathrm{RS}, 7 \mathrm{SR}, 9 a \mathrm{SR}, 9 b \mathrm{SR}$ )-Perhydro-3a,6,6,7,9a-pentamethylnaphtho[2,1-b]furan (12a-f), ( 2 RS/SR, $3 a$ RS, $5 a$ SR, $9 a$ SR, $9 b$ RS) - and (2RS/SR, $3 a$ RS, $5 a$ SR, $9 a$ SR, $9 b$ SR , -Perhydro- $2,3 a, 6,6,9 a-p e n-$ tamethylnaphtho[2,1-b]furan (13a,b), (3aRS,5aSR,9aSR,9bRS)- and (3aRS,5aSR,9aSR,9bSR)-Perhydro-2.2,3a,6,6,9a-hexamethylnaphtho [ 2,1 -b ]furan ( $14 \mathrm{a}, \mathrm{b}$ ). A soln. of the alcohol ( 1 mmol ) in 2-nitropropane ( 5 ml ) was added dropwise, within 10 min to a stirred mixture of $\mathrm{FSO}_{3} \mathrm{H}$ (Bayer, 10 mmol ) and 2 -nitropropane ( 10 ml ) at $-90^{\circ}$ (cooling bath: liquid $\mathrm{N}_{2} / \mathrm{MeOH}$ ) under $\mathrm{N}_{2}$. After the addition, the mixture was poured, with stirring, into sat. aq. $\mathrm{NaHCO}_{3}$ soln. $(50 \mathrm{ml})$ at $0-5^{\circ}$ and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The combined org. phase was washed with sat. aq. NaCl soln. and dried (anh. $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ). Filtration, concentration, and Kugelrohr distillation i.v. (120-140\%/0.04 Torr) afforded the product mixture whose distribution of tricyclic ethers $\mathbf{1 2 a - f}, \mathbf{1 3 a}, \mathbf{b}$, and $\mathbf{1 4 a}, \mathbf{b}$ is presented in Tables 1 and 2. Prep. GC allowed the isolation of stereochemically pure samples of 12a-f whose structures were assigned by comparison (GC/MS and NMR) with analogous compounds [8] [13] ${ }^{1!}$ ). For 13a,b and 14a,b, the unseparated diastereoisomeric mixtures were analysed by comparison with authentic samples [3] or analogues [4].

Data of 12a: ${ }^{1} \mathrm{H}-\mathrm{NMR}: ~ 0.67(s, 3 \mathrm{H}) ; 0.81(s, 3 \mathrm{H}) ; 0.85(d, J=7,3 \mathrm{H}) ; 0.90(s, 3 \mathrm{H}) ; 1.09(\mathrm{~s}, 3 \mathrm{H}) ; 3.86(\mathrm{~m}$, 2 H). ${ }^{13}$ C-NMR: see Table 4. MS: $250\left(2, M^{+}\right), 235(100), 151(33), 137(42), 109(20), 97(47), 81$ (13), 67 (14), 55 (16), 43 (21).

Data of 12b: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.66(s, 3 \mathrm{H}) ; 0.84(d, J=7,3 \mathrm{H}) ; 0.91(\mathrm{~s}, 3 \mathrm{H}) ; 1.07(\mathrm{~s}, 3 \mathrm{H}) ; 1.37(\mathrm{~s}, 3 \mathrm{H}) ; 3.81(\mathrm{~m}$, 2 H). ${ }^{13} \mathrm{C}-$ NMR: see Table 4. MS: $250\left(7, M^{+}\right), 235(47), 151$ (100), 137 (53), 123 (17), 109 (22), 95 (27), 81 (17), 67 (17), 55 (17), 43 (25).

Data of 12c: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.81(d, J=7,3 \mathrm{H}) ; 0.93(\mathrm{~s}, 3 \mathrm{H}) ; 1.04(\mathrm{~s}, 3 \mathrm{H}) ; 1.14(\mathrm{~s}, 3 \mathrm{H}) ; 1.15(\mathrm{~s}, 2 \mathrm{H}) ; 3.85(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13} \mathrm{C}$-NMR: see Table 4. MS: $250\left(1, M^{+}\right), 235(88), 151(22), 137(21), 121(25), 109(29), 97(100), 81(26), 67(30)$, 55 (42), 43 (42).

Data of 12d: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.79(d, J=7,3 \mathrm{H}) ; 0.92(\mathrm{~s}, 3 \mathrm{H}) ; 0.98(\mathrm{~s}, 3 \mathrm{H}) ; 1.09(\mathrm{~s}, 3 \mathrm{H}) ; 1.35(\mathrm{~s}, 3 \mathrm{H}) ; 3.76(\mathrm{~m}$, 2 H). ${ }^{13}$ C-NMR: see Table 4. MS: $250\left(3, M^{+}\right), 235(100), 151$ (27), 137 (28), 123 (31), 109 (31), 97 (89), 81 (42), 67 (37), 55 (50), 43 (76).

Data of 12e: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.84(d, J=7,3 \mathrm{H}) ; 0.97(\mathrm{~s}, 3 \mathrm{H}) ; 0.98(\mathrm{~s}, 3 \mathrm{H}) ; 1.05(\mathrm{~s}, 3 \mathrm{H}) ; 1.14(\mathrm{~s}, 3 \mathrm{H}) ; 3.85(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13}$ C-NMR: see Table 4. MS: $250\left(3, M^{+}\right), 235(100), 151(29), 137(25), 123$ (30), 109 (30), 97 (77), 83 (36), 67 (33), 55 (52), 43 (55).

Data of 12f: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.76(d, J=7,3 \mathrm{H}) ; 0.94(\mathrm{~s}, 3 \mathrm{H}) ; 0.98(\mathrm{~s}, 3 \mathrm{H}) ; 1.02(\mathrm{~s}, 3 \mathrm{H}) ; 1.06(\mathrm{~s}, 3 \mathrm{H}) ; 3.77(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13}$ C-NMR: see Table 4. MS: $250\left(0, M^{+}\right), 235(89), 151(17), 135(14), 121(21), 109(26), 97(100), 81$ (41), 69 (34), 55 (43), 43 (47).

Data of 13a: see [3].
Data of 13b (unseparated 1.2:1 diastereoisomeric mixture at C(2)): Major Isomer: ${ }^{\mathbf{H}} \mathrm{H}$-NMR: $0.81(\mathrm{~s}, 3 \mathrm{H})$; $0.89(s, 3 H) ; 1.09(s, 3 H) ; 1.16(d, J=7,3 H) ; 1.37(s, 3 H) ; 4.11(m, 1 H) . \mathrm{MS}: 250\left(6, M^{+}\right), 235(85), 151(30), 137$ (100), 111 (57), $95(58), 81(60), 43(98)$. Minor Isomer: ${ }^{\text {t }} \mathrm{H}-\mathrm{NMR} ; 0.81(s, 3 \mathrm{H}) ; 0.89(s, 3 \mathrm{H}) ; 1.09(s, 3 \mathrm{H}) ; 1,26$ $(d, J=7,3 \mathrm{H}) ; 1.37(s, 3 \mathrm{H}) ; 4.04(\mathrm{~m}, 1 \mathrm{H}) . \mathrm{MS}: 250\left(8, M^{+}\right), 235(46), 151(46), 137(95), 109(52), 95(48), 81(58)$, 43 (100).

Data of 14a [4]: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.83(s, 3 \mathrm{H}) ; 0.85(\mathrm{~s}, 3 \mathrm{H}) ; 0.87(\mathrm{~s}, 3 \mathrm{H}) ; 1.16(\mathrm{~s}, 3 \mathrm{H}) ; 1.19(\mathrm{~s}, 3 \mathrm{H}) ; 1.35(\mathrm{~s}, 3 \mathrm{H})$. ${ }^{13} \mathrm{C}$-NMR: $81.3(\mathrm{~s}) ; 79.1(\mathrm{~s}) ; 60.4(\mathrm{~d}) ; 57.3(\mathrm{~d}) ; 42.6(t) ; 41.1(t) ; 40.1(t) ; 36.1(\mathrm{~s}) ; 36.0(t) ; 33.6(q) ; 33.1(\mathrm{~s}) ; 30.8$

[^6](q); $24.0(q) ; 21.1(q) ; 20.8(t) ; 18.5(t) ; 15.5(q)$. MS: $264\left(0.5, M^{+}\right), 249(58), 191(35), 137(33), 109(53), 95(52)$, 43 (100).

Data of 14b: ${ }^{1} \mathrm{H}-\mathrm{NMR}: 0.81(s, 3 \mathrm{H}) ; 0.90(s, 3 \mathrm{H}) ; 1.10(s, 3 \mathrm{H}) ; 1.17(s, 3 \mathrm{H}) ; 1.32(s, 3 \mathrm{H}) ; 1.36(s, 3 \mathrm{H})$. ${ }^{13}$ C-NMR: $81.6(s) ; 77.5(s) ; 59.1(d) ; 47.3(d) ; 42.4(t) ; 41.7(t) ; 39.0(t) ; 38.9(t) ; 35.9(s) ; 33.6(q) ; 33.0(s) ;$ $31.2(q) ; 30.8(q) ; 29.5(q) ; 23.0(q) ; 21.9(q) ; 20.7(t) ; 18.6(t) . \mathrm{MS}: 264\left(1, M^{+}\right), 249(52), 191(62), 137(42)$, 109 (53), 95 (44). 43 (100).

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[^0]:    ${ }^{1}$ ) For a review of acid-mediated cyclisations involving $\mathrm{C}-\mathrm{C}$ bond formation, see [2].

[^1]:    ${ }^{2}$ ) Both enantiomerically pure diastereoisomers of ( - )-13a have been synthesised and reported to exhibit a 'typical Ambrox ${ }^{19}$ odour', see [3]; $(-)$-14a has been described as odourless [4].
    $\left.{ }^{3}\right)(-)-3 \mathbf{a}\left(\right.$ Ambrox ${ }^{\mathbb{R}}$; trade name of Firmenich $\left.S A\right)$ is a commercially important odourant naturally occurring in ambergris [5]; for recent syntheses of ( - )-3a and ( $\pm$ )-3a, see [6] and [7], respectively.
    $\left.{ }^{4}\right)(-)-\mathbf{3 b}$ is reported as exhibiting an odour strength unequalled by its diastereoisomers [8]; for a recent total synthesis, see [ 9 ].
    ${ }^{5}$ ) Purchased from Givaudan-Roure, see Exper. Part for analytical details.

[^2]:    ${ }^{6}$ ) All chiral compounds synthesised in this work are racemic.

[^3]:    ${ }^{7}$ ) Strong evidence for this assumption is provided by MM2 molecular-mechanics calculations, ${ }^{13} \mathrm{C}$-NMR data (see Table 3), and by analogy with known work [12].
    ${ }^{8}$ ) Products derived from transition states having a skew-boat conformation of the nascent cyclohexane ring were also postulated in our previous work [1].
    ${ }^{9}$ ) A slight stereochemical bias (ca. 1.2:1 diastereoisomeric mixtures) favours the transition state with a pseudoequatorial Me group in the developing tetrahydrofuran ring (i.e. for 13a, the diastereoisomer with an $\alpha-\mathrm{Me}-\mathrm{C}(2)$ group is preponderant, whereas for 13 b the favoured diastereoisomer has a $\beta-\mathrm{Me}-\mathrm{C}(2)$ group; identification was effected by GC/MS and NMR comparison with authentic samples [3]).

[^4]:    ${ }^{10}$ ) In our earlier work [1], we postulated that isomerisation of this double bond may be due to neighbouring group participation of the protonated homoallylic OH group.

[^5]:    ${ }^{\text {a }}$ ) COSY, C,H correlation. ${ }^{\text {b }}$ ) Interchangeable.

[^6]:    ${ }^{11}$ ) The preferred conformation of $c i s$-fused $A / B$ ring diastereoisomers 12c-f is indicated by their NMR data and molecular-mechanics calculations using the MACROMODEL program [15]. In analogy to the previously studied $C(7)$-demethylated tricyclic ethers [13], we believe that the chair/chair conformation, in which the $\mathrm{Me}-\mathrm{C}(7)$ group is equatorial and the $\mathrm{Me}-\mathrm{C}(9 \mathrm{a})$ group is axial in ring $A$, is preferred for $\mathbf{1 2 c}, \mathbf{1 2 d}$, and $\mathbf{1 2 f}$. For 12e, however, the equatorial $\mathrm{Me}-\mathrm{C}(7)$ group indicated by the NMR data implies a chair/skew-boat conformation for rings $A$ and $B$, in which the $\mathrm{Me}-\mathrm{C}(9 \mathrm{a})$ group is axial to the ring $A$. In the alternative chair/chair conformation, the $\mathrm{Me}-\mathrm{C}(7)$ group is axial and the $\mathrm{Me}-\mathrm{C}(9 \mathrm{a})$ group equatorial. It is gratifying to note that calculations indeed favour the former conformation, albeit by only $0.3 \mathrm{kcal} / \mathrm{mol}$. For the $\mathrm{C}(7)$-demethylated analogue of $\mathbf{1 2 e}$ [13], the NMR data show that this preference is reversed, again in agreement with calculations which, in this case, estimate the chair/chair conformation to be $1.1 \mathrm{kcal} / \mathrm{mol}$ lower in energy than the chair/skew-boat conformation.

